

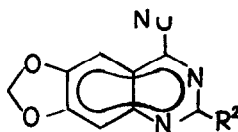
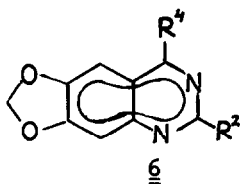
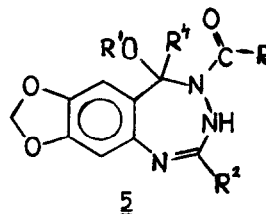
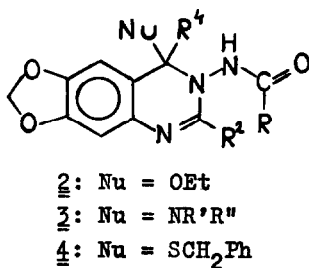
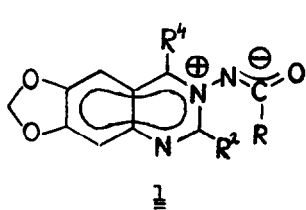
ELECTRON DEFICIENT HETEROAROMATIC AMMONIOAMIDATES, X¹
PHOTOLYSES OF SOME N-(3-QUINAZOLINIO)-AMIDATES²

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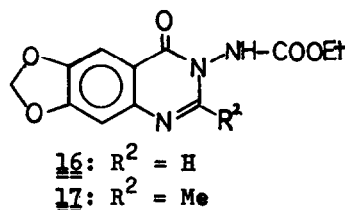
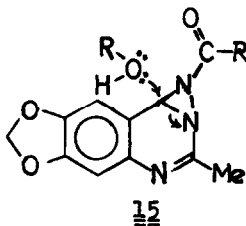
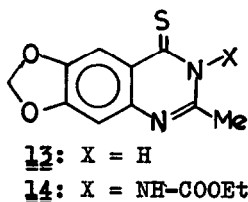
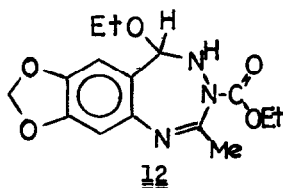
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The N-(3-quinazolinio)-amidates 1a-e do exist in alcoholic solns as equilibrium mixtures with their alcohol adducts, e.g. 2a-e.³ When an Ar purged ethanolic soln of 1a was irradiated through Pyrex with a high pressure Hg lamp, four products,⁴ 5a (R'=Et, 26 %, mp. 174-5°), 6a (30 %, mp. 180°), 7a (5 %, mp. 126-8°) and ethyl carbamate, were obtained. 6a and 7a were identical with authentic samples. The structure assignment of 5a (R'=Et) is based on the following observations. UV [209 (4.56); 220 (4.4), sh; 300 (4.00) in EtOH], IR [KBr; ν_{NH} 3300, b, $\nu_{\text{C=O}}$ 1690] and mass spectra [m/e 321: (M) 44 %, m/e 276: (M[±] OEt)⁵ 100 %, m/e 248: (276[±] C₂H₄) 15 %, m/e 230: (276[±] EtOH) 20 %, m/e 204: (248[±] CO₂) 9 %, m/e 189: (C₁₀H₉N₂O₂) 28 %, m/e 188: (230[±] NCO) 16 %] are consistent with the isomeric structures 5a (R'=Et) and 12; the latter, however, is ruled out by the PMR spectrum [OEt: δ 1.28, t + 3.6, q, J = 7 Hz; COOEt: 1.35, t + 4.3, q, J = 7 Hz; 2-Me: 2.15, s; OCH₂O: 5.95, s; 5-H + ArH's: 6.35 + 6.4 + 6.7, all s; NH: 7.1, bs in CDCl₃; OEt: 1.08, t + \sim 3.45, m, J = 7 Hz; COOEt: 1.14, t + 4.10, q, J = 7 Hz; 2-Me: 1.96, s; OCH₂O: 6.01, s; 5-H + ArH's: 6.35 + 6.6 + 6.9, all s; NH: 8.8 bs in DMSO-d₆] because there is no H-N-C-H coupling as required by structure 12, while such a coupling ($J_{\text{H-C}(2)\text{-N-H}} = 5$ Hz) is observed in the DMSO-d₆ PMR spectrum of the analogous 5d (R'=Et). When irradiation of 1a was performed in benzyl alcohol, 5a (R'= PhCH₂, 4 %, m.p. 90-3°) was formed along with 6a and ethyl urethane.



<u>1</u> - <u>11</u>			
	R ²	R ⁴	R
<u>1b</u>	Me	H	OEt
<u>1c</u>	Me	H	OCH ₂ Ph
<u>1d</u>	Me	H	Ph
<u>1e</u>	H	H	OEt
<u>1f</u>	Me	Me	OEt



Irradiations of 1b-d in EtOH furnished the analogous benzotriazepines 5b (R'=Et; 23 %, mp. 176°), 5c (R'=Et; 26 %, mp. 197°) and 5d (R'=Et; 11 %, mp. 122-4°), respectively, whose UV, IR, mass and PMR spectra are in agreement with their assumed structures. Irradiation of 1e, for which the equilibrium with its EtOH adduct is shifted towards the amidate,³ furnished 6e (86 %, m.p. 181-3°).

When benzylamine or α -toluenethiol were added to CDCl₃ solutions of 1c, the signals corresponding to 1c³ were gradually replaced by those of the

adducts 3c (NRR' = NHCH₂Ph; 2-Me: δ 2.25; 4-H: 4.8; OCH₂O: 5.75; ArH's: 6.8 + 7.1)⁶ and 4c (OCH₂O: δ 5.90), respectively.

Irradiation of 1a in BuNH₂, PhCH₂NH₂ and morpholine furnished 8a (70 %, mp. 193°), 9a (42 %, mp. 199-200°) and 10a (70 %, m.p. 187°), respectively, whose structures were derived from their IR and mass spectra, and corroborated by comparison with authentic samples. Irradiation of 1e in BuNH₂ furnished 6e (80 %). Ethyl urethane was obtained as the by-product in all cases.

When 1a was irradiated in a mixture of CHCl₃ and α -toluenethiol, 11a (4 %, mp. 170-1°), 13 (4 %, mp. 290-2°) and 14 (25 %, mp. 170-1°) were obtained as the major products. 11a and 13 proved identical with authentic samples, while structure 14 was derived from the mass spectrum [m/e 307: (M) 100 %, m/e 274: (M - SH) 12 %, m/e 262: (M - EtO) 11 %, m/e 235: (M - CO₂ - C₂H₄) 16 %, m/e 220 (262 - NCO) 20 %, m/e 206 (235 - 29) 9 %, m/e 205: 9 %, m/e 187: (220 - SH) 17 %], IR [ν NH 3240, ν C=O 1740 in KBr] and the PMR spectrum [NH: δ 8.1, bs; 5-H: 7.98, s; 8-H: 7.05, s; OCH₂O: 6.18, s; 2-Me: 2.68, s; COOEt: 1.35, t + ~4.3, m, J = 7.3 Hz, in CDCl₃] of the product.

The Type 7-11 compounds are thought to arise from a Norrish Type II cleavage of the Type 2-4 adducts,⁷ respectively, while the Type 6 compounds are clearly the acylnitrene elimination products of the amidates 1 themselves, and probably formed through the T₁ states (c.f. Ref. 8). The ring enlargement products 2, on the other hand, are probably formed via the S₁ states and the diaziridine intermediates 15. The formation of 6e as the only heterocyclic product on photolysis of 1e in EtOH and BuNH₂ reflects the reduced propensity of C-4 of the latter towards being attacked by nucleophiles. 11a, the primary photolysis product of 5a formed in the presence of α -toluenethiol is, as shown in a separate experiment, photounstable. Photolysis of its S-CH₂ bond leads to 13. Formation of 14, on the other hand, may be rationalized by assuming photolysis of the S-CH₂ bond of 4a and H abstraction by the resulting heterocyclic radical.

An interesting by-product of the photolysis of 1d in EtOH is 16 (10 %, m.p. 226°), whose 2-Me derivative 17 (13 %, mp. 160-2°) is one of the products

formed on prolonged refluxing of 1a with BuNH₂. The structure of 17 was proven by comparison with an authentic sample, while that of 16 follows from the similarity of its mass, IR, UV and PMR spectra to those of 17.

References and Notes

1. Part IX: B. Ágai and K. Lempert, submitted for publication
2. Part II of N-(3-Quinazolinio)amidates. For Part I, see Ref. 3.
3. J. Fetter, K. Lempert and J. Møller, Tetrahedron 31 (1975), accepted for publication
4. All new compounds gave satisfactory microanalyses
5. An asterisk indicates that the decomposition is accompanied by the appropriate metastable peak
6. For the formation of an analogous NH₃ adduct, see Ref. 3.
7. The dihydroderivative 2a (with H instead of Nu) similarly yields 6a, when irradiated in EtOH.
8. J. Streith, T.P. Luttringer and M. Nastasi, J. Org. Chem. 36, 2962 (1971)